"Soft Materials Design by Computer Simulations"

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NORTHWESTERN UNIVERSITY Soft materials constitute the basic component of living systems, are integrated into the fabric of modern society, and will play a key role in futuristic devices.



Examples:

Liquid crystals, colloids, polymers, surfactants, membranes (liquid or elastic), gels, viruses,...







Soft Materials are Hard

- Polymers and soft materials are for the most part amorphous fluids or solids with embedded structures spanning nanometers to microns and even millimeters or more.
- Soft materials can possess an extraordinarily broad spectrum of *timescales* often covering more than ten orders of magnitude.
- In many cases the relaxation processes contributing to this broad spectrum involve nonequilibrium phenomena such as vitrification, jamming, or constrained crystallization

Motivation

- Because of the combination of chemical versatility and the ability to create structures with unique tailored features at different length scales, soft matter entities are attractive materials from which to construct novel materials
- Soft matter is fascinating and there are many open problems
- Advances rely on physics, chemistry, biology and engineering

Shells in the Living World



Viral capsids are the shells of viruses that contain DNA or RNA) are icosahedra when they are large



Organels are cellular nano-compartments (nano-reactors that synthesize energy) that take many polyhedral geometries (Bacterial microcompartments Fan CG, et al. PNAS (2010)

Dodecahedral Coccolithophorids





0.3 µт

Glinel,K. et al.

1 µm

Braarudosphaera bigelowii Braarudosphaera bigelowii

images from Jeremy Young's web page National History Museum, London, UK images courtesy of Pupa Gilbert, UW-Madison



Viruses (ex: Mimivirus)

Biological examples include viruses, organelles and archaea organisms



(T.O. Yeates et al Nature Rev. 2008)

(lancu, et al., J. Mol. Biol. (2010))



Carboxysomes are bacterial microcompartments (~80 to 140 nm) that contain enzymes (like RuBisCO) involved in carbon fixation.

These organelles are found in all bacteria that fix carbon dioxide.



Bacterial microcompartments Fan CG, et al. PNAS (2010) Halophilic organisms live at very high salt concentrations

ARCHAE organisms envelops take polyhedral shapes. They have hexagonal packing crystalline surfaces



Walby's Square Baterium (W. Stoeckenius J. BACTERIOLOGY, (1981), 148, 352-360 . Has a hexagonal lattice (W. Stoeckenius J. BACTERIOL. (1981))



Ionic Membranes What is the shape of crystalline ionic shells?

palmitic (water insoulble) acid trilysine (micelle in water)



X-ray scattering and TEM show a hexagonal lattice that buckled into low symmetry shapes: WHY?

These crystalline shells, are stable even at high salt concentrations. Good for biomedical applications.



3;1 cationic-anionic amphiphiles (Greenfield, et al., *JACS 2009; Leung et al ACS Nano, 2012)*



Why the icosahedron is ubiquitous?





Defects on curved surfaces

g=1

 $N_5 = N_7 = 0$

 $N_5 - N_7 = 12(1 - g)$



 $N_5 = 12, N_7 = 0$

Defects (disclinations) on curved crystalline closed spherical surfaces



Defects are in icosahedral positions (Bowick, Nelson, Travesset Phys. Rev. B (2000))

Icosahedral symmetry is not icosahedral buckling (Lidmar, et al, Phys. Rev. E (2003))



 $Y - Young's modulus, \kappa$ - bending rigidity

On a sphere all 12 defects simultaneously act as seeds for buckling.



Lidmar, et al, PRE (2003)

How about heterogeneities?

* of elastic parameters

Elastically heterogeneous shells



- Shells made by more than one protein or components
- They for many polyhedral geometries with patterns even without chemical incompatibility.
- That is, even compatible components are segregated due to their different mechanical properties which leads to specific buckled shapes.
- The patterns are coupled to the shape of the shell.

Structures of shells

 $\kappa_{hard}/\kappa_{soft}=50, Y_{hard}/Y_{soft}=1$



(G. Vernizzi, R. Sknepnek, M. Olvera de la Cruz – PNAS 108, 4292 (2011))

prefers to buckle prefers to stay flat



1-hosohedron and 2-hosohedron (top row, on the left), regular platonic solids (tetrahedron, cube, octahedron, dodecahedron and icosahedron) and the truncated polyhedra. Fragmentation!!



prefers to buckle

prefers to stay flat

Salmonella Pdu Microcompartments, courtesy of Danielle Tullman-Ercek, UCB (note, when only one protein is present shape is spherical: Schmidt-Dannert Plos One 7, e33342 (2012))



Experiments

palmitic acid trilysine

+3 cation from three lysine amino acid residues



palmitic acid -1 anion from carboxylic acid



Mixtures of 3+ and -1 amphiphiles





WAXS region



- Lattice was formed only with cation + anion mixture
- The scan with cation micelles alone do not have sharp peaks
- Electrostatic attraction induce crystallization of tails

(Leung et al ACS Nano, 2012)

SAXS region

5.3nm

Catanionic bilayer at pH~3 (only some anion ionized)



3.9 nm (hydrophobic layer)

Analog to

30% ionization of palmitic acid in atomistic simulation



Catanionic bilayer at pH~8 (anion fully ionized)



(Leung et al ACS Nano, 2012)

Analog to 95% ionization of palmitic acid





Hard and thick

Thinner membrane (h_s) and more disorder along the highly bended areas than along the flat areas (h_h) . Mechanics of thin solids suggests that the bending rigidity goes as h^3 (so is very less sensitive to thickness)

pH=5, hexagonal lattice

For R=50nm vesicle with $E_s \sim \alpha(f)R^2$ 25x25(=625)nm² crystalline domains (from WAXS broadening of peaks), there are ~ 50 platelets. Small R gives less symmetric vesicles than large R (more spherical).

STABLE AT HIGH SALT





 $= 0.197 \text{ nm}^2$

 $= 0.197 \text{ nm}^2$

= 0.205 nm²

 $= 0.197 \text{ nm}^2$

 $= 0.197 \text{ nm}^2$

Phase diagram for 2D crystal lattice 2+/-1 amphiphiles (tail length versus pH)

(Leung et al, PNAS 2013)



Small window on chain length where all interactions are of the same order of magnitude KT give pH re-entrant transitions

Analogy to ceramics

Ceramics are typically 3D crystals with ionic bonds.

Ionic square crystals break up due to curvature.



Ceramics are strong until they shatter under shear.



from wikipedia

Summary

•Regular and irregular polyhedral geometries, arise spontaneously in shells formed by more than one component such as cellular containers.

•While spherical and icosahedral shells are abundant in nature, the less symmetric polyhedral shells are potentially more functional and appear at the nanometer length scales where their functionality can be fully exploited.

•Ionic closed crystalline membranes have hexagonal lattices (important for design).

•A subtle balance of interactions allows transitions by external parameters (pH in this case).

Ionic nanocontainers experiments and modeling over various length scales



Our model predicts faceting into many possible polyhedra with decorated surfaces that offer higher functionality





Strongly charged polyelectrolyte gels expand in monovalent salts and adsorb water (dipper) but collapse in divalent salts (chromosomes)





Monovalent swollen gel Divalent compacted gel

Soft Matter; JCP; PRE; Macromolecules

Physics: Swell to increase osmotic pressure of the ions and collapse due toionic correlations Challenge: Structures are not periodic like molten salts!

Supramolecular materials mimic biological fibers or cytoskeleton of cells including microtubules, actin filaments, and intermediate filaments and tissues offering the possibility to fabricate materials with the properties of living systems.







Hydrogels as synthetic cell scaffolds





Substrate stiffness influences cellular

adhesion

proliferation

movement

differentiation

GOAL: Develop methods to align peptide amphiphile (PA) nanofibers.

Cardiac muscle in the heart¹

Neurons in the brain²

Highly aligned cellular environments:

- smooth muscle
- brain
- cardiac muscle
- spinal cord

Challenge: Alling micron size fibers that are soft and small to be manipulated directly

micrometers long with diameter ~10 nm

Milimmeter thread





Nature Materials 9, 594 (2010)

Preferential alignment of human mesenchymal stem cells along the axis of a noodle

Action potential propagation in a noodle seeded with cardiomyocytes Conductive noodle with carbon nanotubes



Images courtesy of S. Zhang and J. Mantei

Heat Induced Alignment Only annealed PA leads to alignment

Cooling: Plaques break into bundles of fused fibers as water molecules rehydrate the PA

nonannealed PA solutions

Noodles with PA solutions **annealed** at 80 °C

















Images courtesy of Dr. A. Mata and Dr. R. Bitton

Peptide Amphiphile-Polymer Hybrid Materials



PA-polymer hybrids contracts at high T

Next steps:

Control of gel

shape and direction of contraction

of such

system

Making aligned gels:



Nat. Mater. 2010, 9, 594







25°C

47°C



Buckling in Heterogeneous Elastomers

Bi-trips gels lead to looped patterns in gut

tubes, which twists as a bi-layer gel

Gut tube and associated sheet tissue





T. Savin et al. Nature 476 (2011), 57-62



Bi-layer gel in poor solvent



R. Erb et al. Nat. Commun. 4 (2013), 1712



Top: Free end. Down: Fixed end



Simple Perversion

A perversion is a kink that connects two helices with opposite chirality observed in natural and artificial mechanical systems.

Perversions provide the fundamental mechanism of helical symmetry breaking and is an actuator.

The elastic energy condenses around the perversion during buckling and at the interface corners



Multiple Perversions and Symmetry Breaking

- Symmetry breaking during buckling
 - Rightmost
 perversion wind
 before the other
 two
- Repulsive interaction between perversions 1 and 2



Soft materials are integral to developments in organic electronics, such as energy harvesting organic photovoltaic devices (OPV) and new display and lighting technologies (OTFT, OLED), and to advanced medical devices and therapies (implants, tissue engineering, drug delivery, personalized medicine).

Molecular electrolytes and ion-containing polymers are the basis for products ranging from super-absorbents, separators, and membranes for advanced energy devices such as batteries and fuel cells.

Polymer electrolytes- Industrial applications



Hallinan Jr., D.T.; Balsara, N.P. Annu. Rev. Mater. Res. 2013. 43:503.

Polymer electrolytes

- Powerful Li-ion batteries can power entire factories
 - Heavy
 - Bulky
 - Expensive
 - Combustible
- Polyelectrolytes
 - Flexible
 - Lightweight
 - Low-cost
 - Recyclable
 - Space applications



Block co-polymer electrolytes

- Multi-scale interactions
- Repulsion of polymer chains (10-100 nm)
- Coupling of opposite charges (1 nm)
- Combine approach that bridges the two length scales (SCFT-LST)







Self-assembly of inverted phases



Dynamics of mobile ions Cathode (+) Anode (-) Charging

- Transport of ions through the polyelectrolyte nanodomain
- High ionic mobility required for high battery performance

Molecular Dynamics



Counterion mobility

$$\mu = \frac{v_{drift}}{qE}$$

Li et al Macromolecules 2016

Coarse-grained Model:

- bead-spring chain (N monomers)
- steric interaction (LJ)
- bond interaction (FENE)
- Coulomb interaction (PPPM)
- implicit solvent

Electrostatics:

- monomers q = +1;
- counterions, q = -1;
- Bjerrum length:

$$l_B = \frac{e^2}{4\pi\epsilon_0\epsilon_r k_B T}$$

Strong non-linear effects



 The mobility increases as the confinement/ concentration increases at low electric field E and it is E independent up to E=E* and or E> E* it increases with E until it saturates at E>>E*, qE*

Opportunities

- Soft materials will play a key role in the technologies that will underpin new business opportunities, employment, and economic vitality since they impact sectors of global energy, water, medicine, agriculture, and sustainability.
- There are many opportunities in energy production, storage and conversion, water purification and reuse, and biotechnology

Opportunities in Theoretical and Computational Polymeric Materials and Soft Matter A National Science Foundation Sponsored Workshop

Chairs: Monica Olvera de la Cruz and Michael Rubinstein Santa Barbara, California, October 20-22, 2013



Topics:

Reversing the Arrow: Materials by Design

Exploiting Geometry in Form and Function

Harnessing Non-Equilibrium Processes

New Paradigms Inspired by Nature

Bridging Scales in Space and Time

Liu et al Soft Matter, 11, 2326-32 (2015)

THANK YOU!